

2015  
**BIOCHEMISTRY**  
**Paper – BCT-303**  
**(Molecular Interaction)**  
**Full Marks – 25**

*The figures in the margin indicate full marks*

*Candidates are required to give their answers in their own words as far as practicable*

Answer **any eight** questions of which **Question no. 1** is compulsory

1. Fill in the blanks :

3.

Tyrosine and tryptophan are significantly more polar than phenylalanine, because of the tyrosine \_\_\_\_\_ and the \_\_\_\_\_ of the tryptophan indole ring. Tryptophan and tyrosine and to a much lesser extent \_\_\_\_\_, absorb ultraviolet light.

The secondary amino (imino) group of \_\_\_\_\_ residues is held in a rigid conformation that reduces the structural flexibility of polypeptide regions containing \_\_\_\_\_.

The polarity of serine and threonine is contributed by the \_\_\_\_\_ groups; that of cysteine by its \_\_\_\_\_ group; and that of asparagine and glutamine by their \_\_\_\_\_ groups.

Cysteine is readily oxidized to form a covalently linked dimeric amino acid called \_\_\_\_\_, in which two \_\_\_\_\_ molecules or residues are joined by a disulfide bond.

2. Fill in the blanks :

3

If a polypeptide chain has a long block of \_\_\_\_\_ residues, this segment of the chain will not form an  $\alpha$  helix at pH 7.0. The negatively charged carboxyl groups of adjacent \_\_\_\_\_ residues repel each other so strongly that they prevent formation of the  $\alpha$  helix. For the same reason, if there are many adjacent \_\_\_\_\_ and/or \_\_\_\_\_ residues, which have positively charged R groups at pH 7.0, they will also repel each other and prevent formation of the  $\alpha$  helix. The bulk and shape of \_\_\_\_\_, Ser, Thr, and Cys residues can also destabilize an  $\alpha$  helix if they are close together in the chain.

**Or**

A \_\_\_\_\_ residue introduces a destabilizing kink in an  $\alpha$  helix. In addition, the \_\_\_\_\_ atom of a Pro residue in peptide linkage has no substituent hydrogen to participate in hydrogen bonds with other residues. For these reasons, \_\_\_\_\_ is only rarely found within an  $\alpha$  helix. \_\_\_\_\_ occurs infrequently in  $\alpha$  helices for a different reason : it has more conformational flexibility than the other amino acid residues. Polymers of \_\_\_\_\_ tend to take up coiled structures quite different from an  $\alpha$  helix.

[Turn Over]

3. The stabilization of the tertiary structures of polypeptide chains is ensured by certain restrictions on the organization of the motifs. Describe them in brief. 3
4. What is the prerequisite condition for reformation of eight disulfide bonds while folding denatured ribonuclease? How did Anfinsen demonstrate that? 3
5. What form of helix is RNA and what are its characteristics? Is the RNA chain a helix or random coil? 3
6. Briefly describe how MC-SYM program proceeds to predict RNA conformation. 3
7. Can the  $\beta$  sheets of proteins and Stem-Loops in RNA be described as secondary structures? If not, explain why? 3
8. Describe how the residues at the interface of a protein-protein complex are identified from the PDB file representing the 3D structures of the protein-protein complex. 3
9. Describe the basic principle of 'Yeast two hybrid system' method for detecting protein-protein interactions. 3
10. Provide a schematic diagram of surface plasmon resonance detection unit. 3
11. Name the factors on which the SPR angle depends. Correlate: Shift in SPR angle and Resonance unit. 3
12. Explain the parameters describing chain initialization and chain elongation. Consider a hypothetical chain element containing four elemental residues A each having either + or - states (+ representing helix and - representing coil). What will be the partition function? 3
13. Consider 30C values for  $\sigma$  and s values for glycine L-Alanine, L-Serine and L-Leucine as given by  $(10^{-5}, 0.62)$ ;  $(8 \times 10^{-4}, 1.06)$ ;  $(7.5 \times 10^{-5}, 0.79)$ ;  $(33 \times 10^{-4}, 1.14)$ . Interpret. 3
14. Sketch out the differences in the mechanisms of binding of a zinc finger protein and a 434 repressor protein with a double stranded DNA. 3
15. Discuss the basic advantages of using microcalorimetry to study DNA-protein interaction. 3
16. You are working with a single stranded RNA virus and found that the terminal RNA sequences are involved in binding with a protein which is important for the viral proliferation. Outline a strategy based on peptidomimetics for developing an antiviral compound in relation to your observation. 3
17. Mention five different methods used for characterising RNA protein interactions. 3

*One mark for neatness*